

Big Bang / The origin of electronics



MIEET

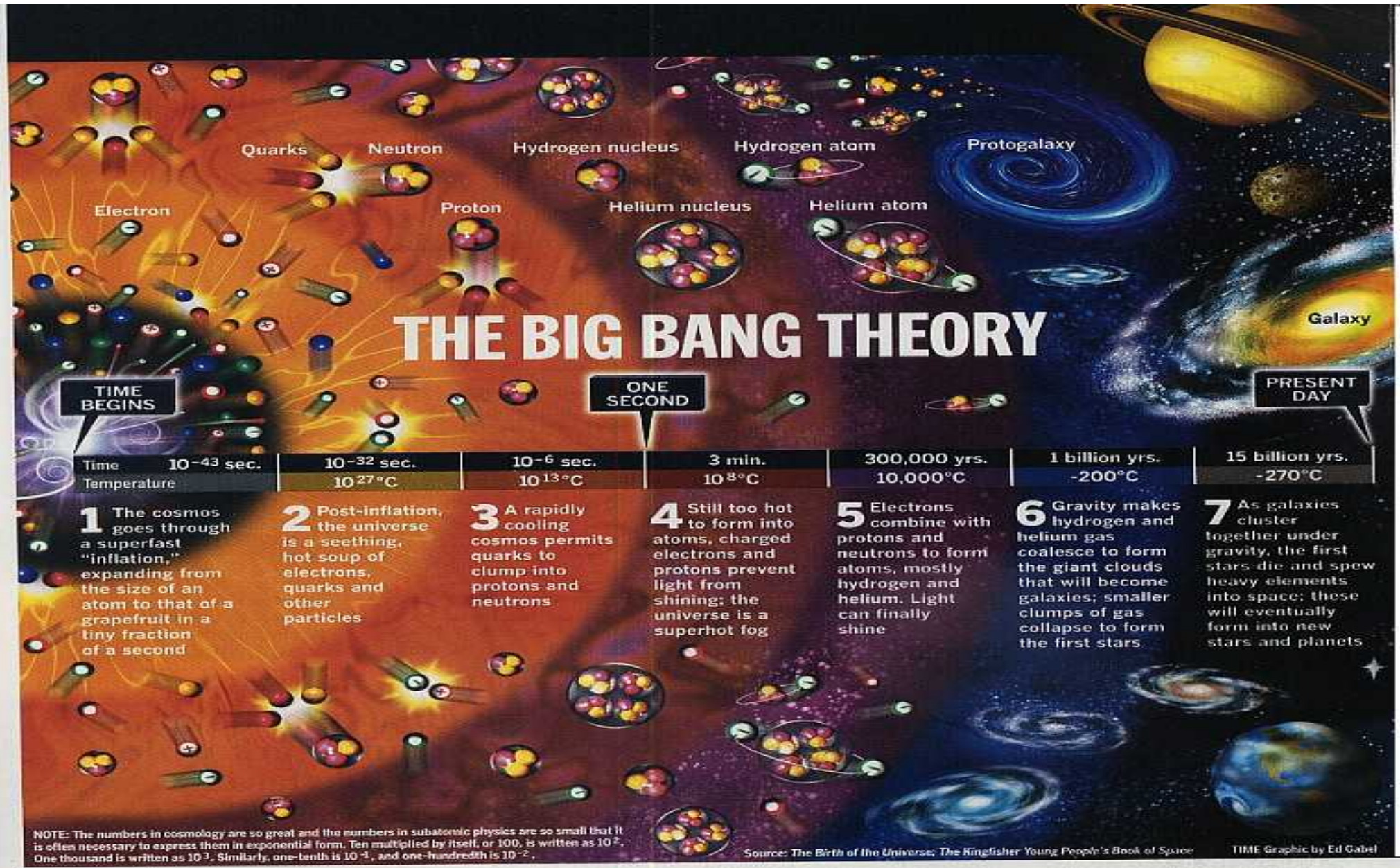
1^o ano



UAAlg
UNIVERSIDADE DO ALGARVE



Big Bang Theory



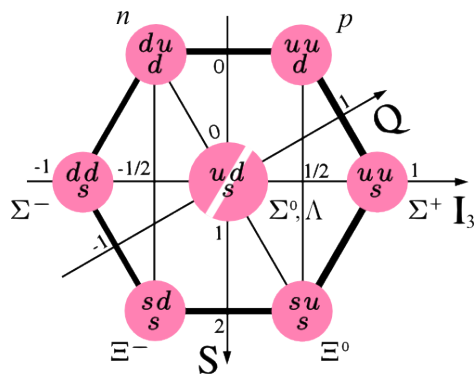
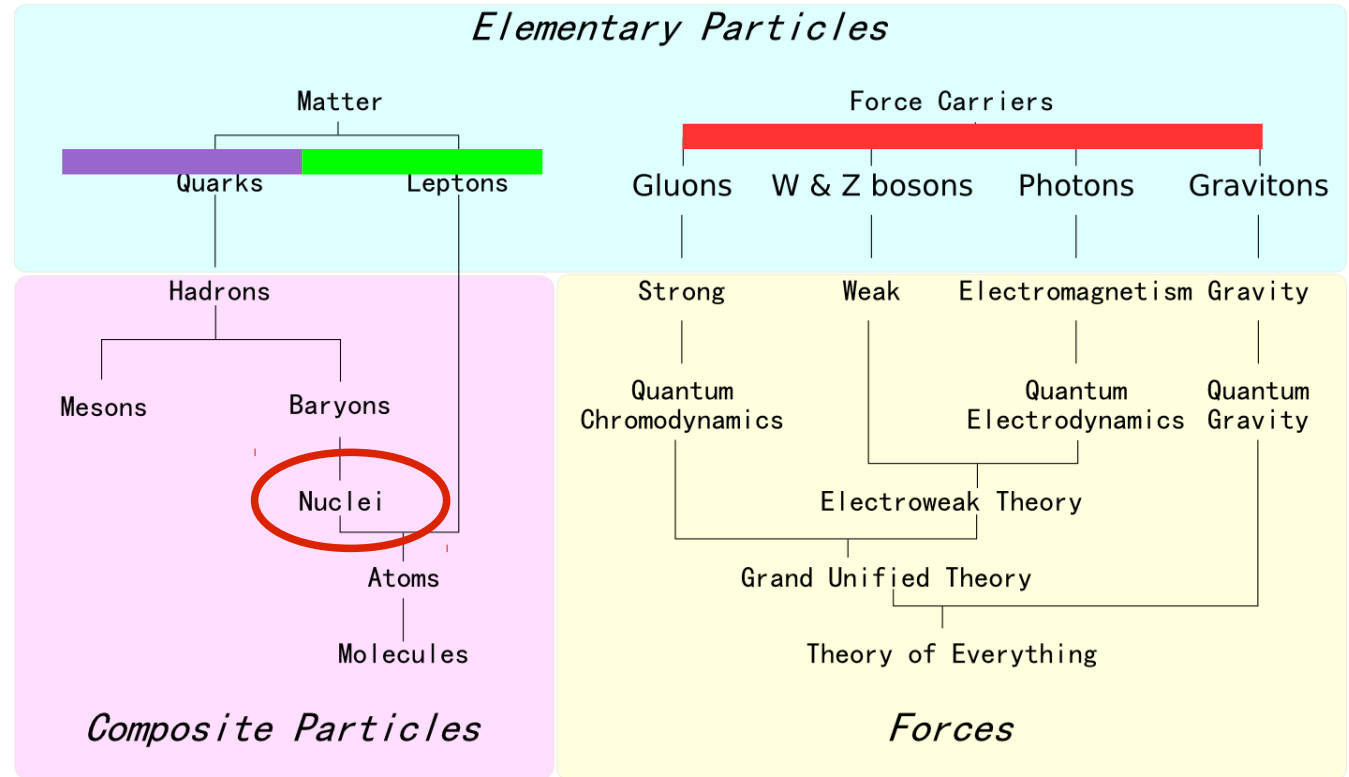
The origin of matter

Everything in the universe has quantum numbers:

- Mass (energy)
- Charge
- Spin
- Baryon number

Three Generations of Matter (Fermions)

	I	II	III	
mass	2.4 MeV	1.27 GeV	171.2 GeV	0
charge	$\frac{2}{3}$	$\frac{2}{3}$	$\frac{2}{3}$	0
spin	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	1
name	u up	c charm	t top	γ photon
	4.8 MeV	104 MeV	4.2 GeV	0
	$-\frac{1}{3}$	$-\frac{1}{3}$	$-\frac{1}{3}$	0
	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	1
Quarks	d down	s strange	b bottom	g gluon
	<2.2 eV	<0.17 MeV	<15.5 MeV	91.2 GeV
	0	0	0	0
	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	1
	ν_e electron neutrino	ν_μ muon neutrino	ν_τ tau neutrino	Z weak force
	0.511 MeV	105.7 MeV	1.777 GeV	80.4 GeV
	-1	-1	-1	± 1
	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$	1
Leptons	e electron	μ muon	τ tau	W weak force

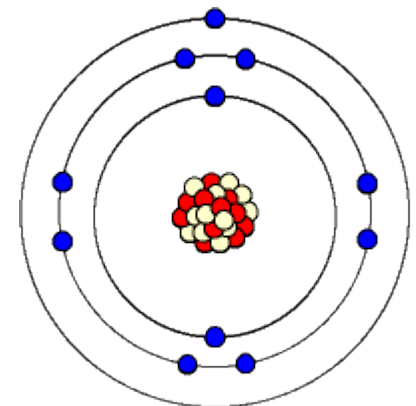


Baryon = 3 quarks

- Neutron (ddu). Charge: $-\frac{1}{3} + -\frac{1}{3} + \frac{2}{3} = 0$
- Proton (uud). Charge: $-\frac{1}{3} + \frac{2}{3} + \frac{2}{3} = 1$

Lepton

- Electron (e). Charge: -1



The origin of matter

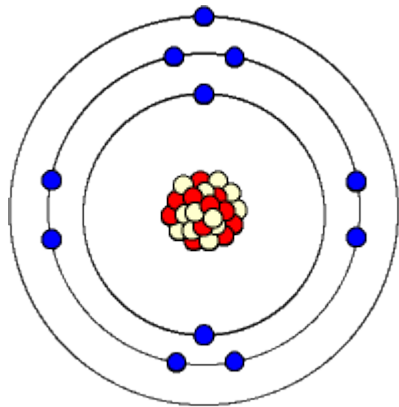
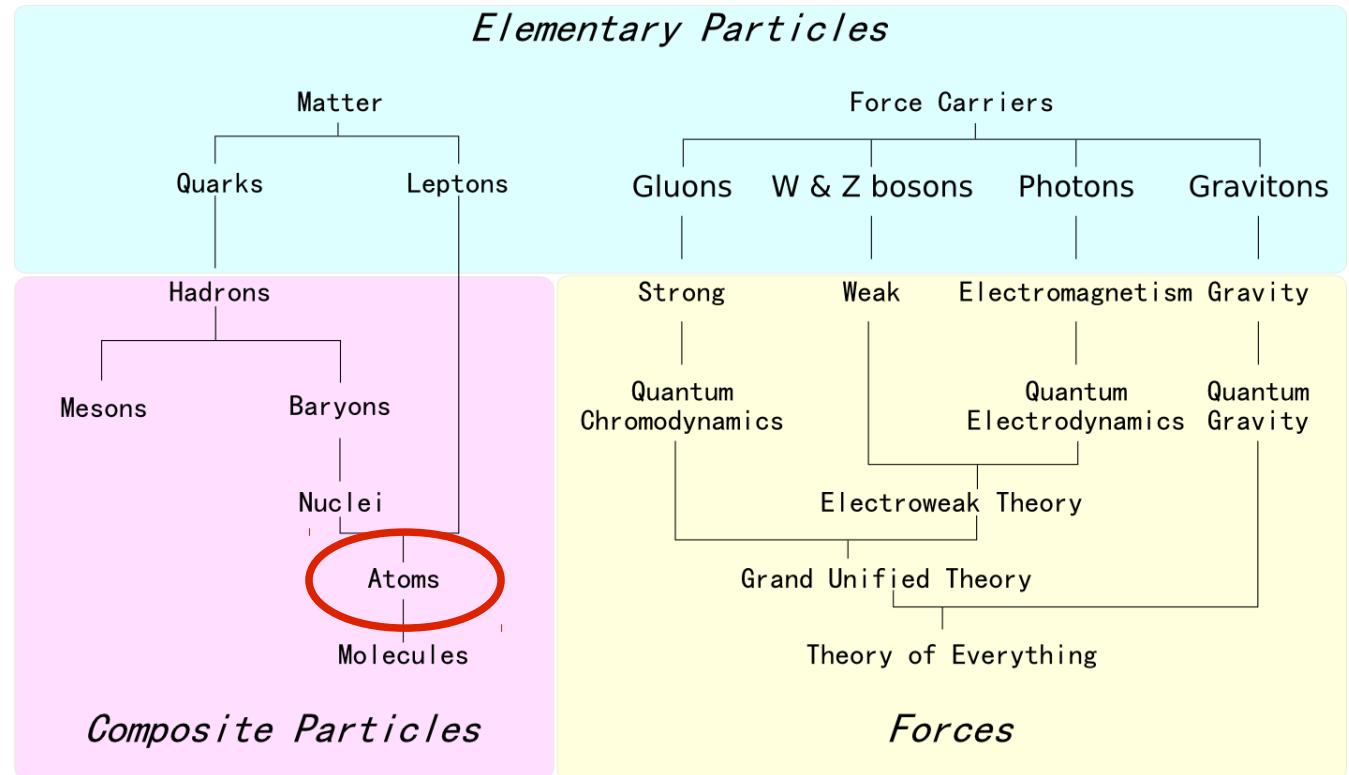
Baryon = 3 quarks

- Neutron (ddu). Charge = 0

- Proton (uud). Charge = 1

Lepton

- Electron (e). Charge: -1



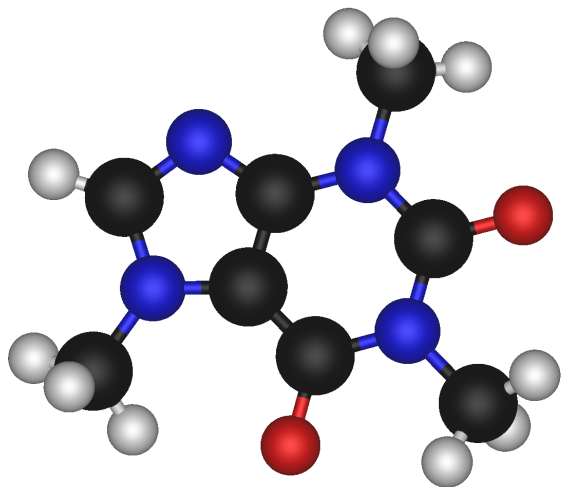
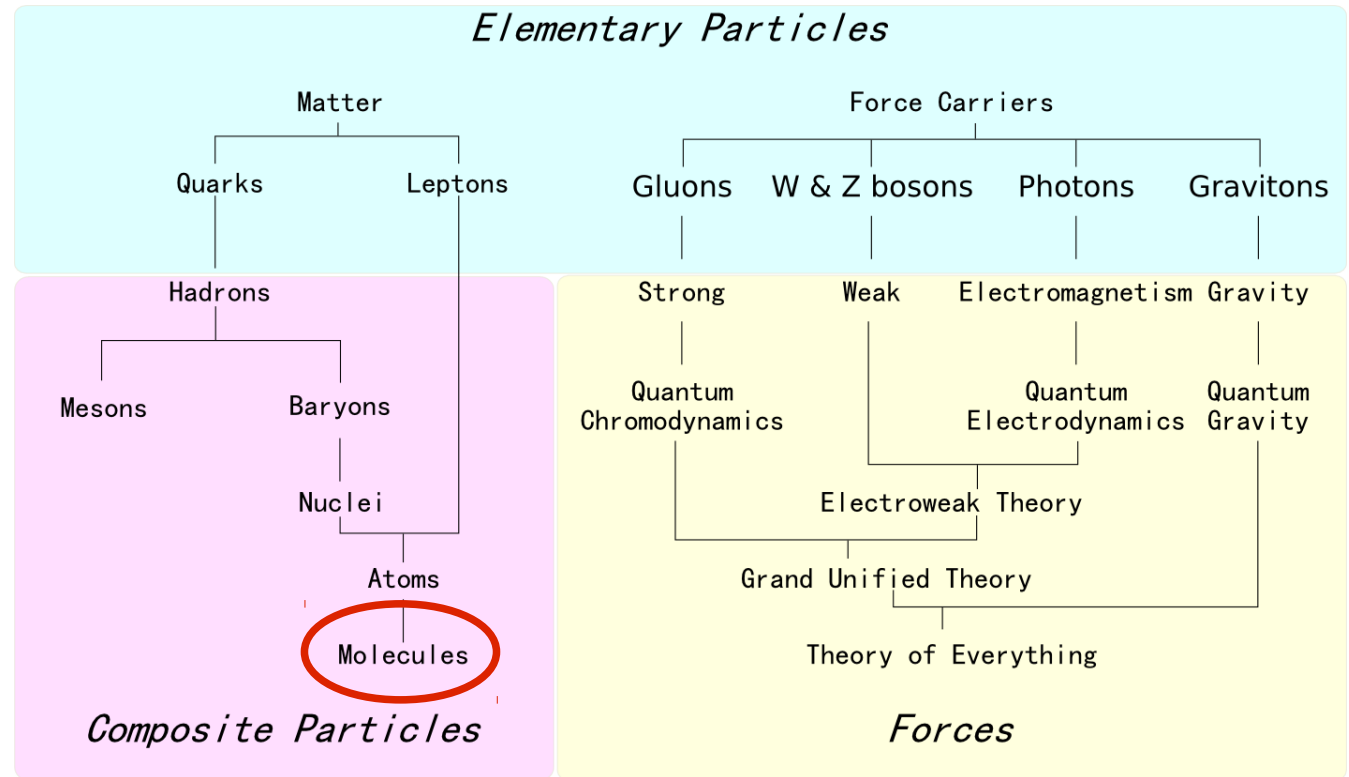
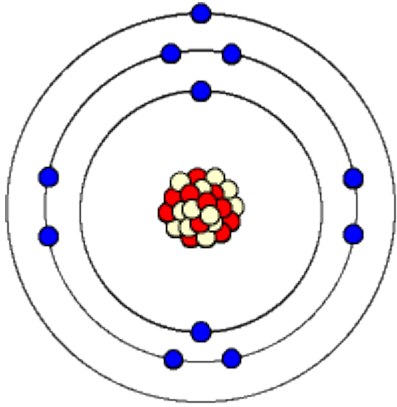
Atom:

n protons + n electrons + m neutrons

Charge = 0

Example: Na (sodium): $n = 11$, $m = 11$ or 12

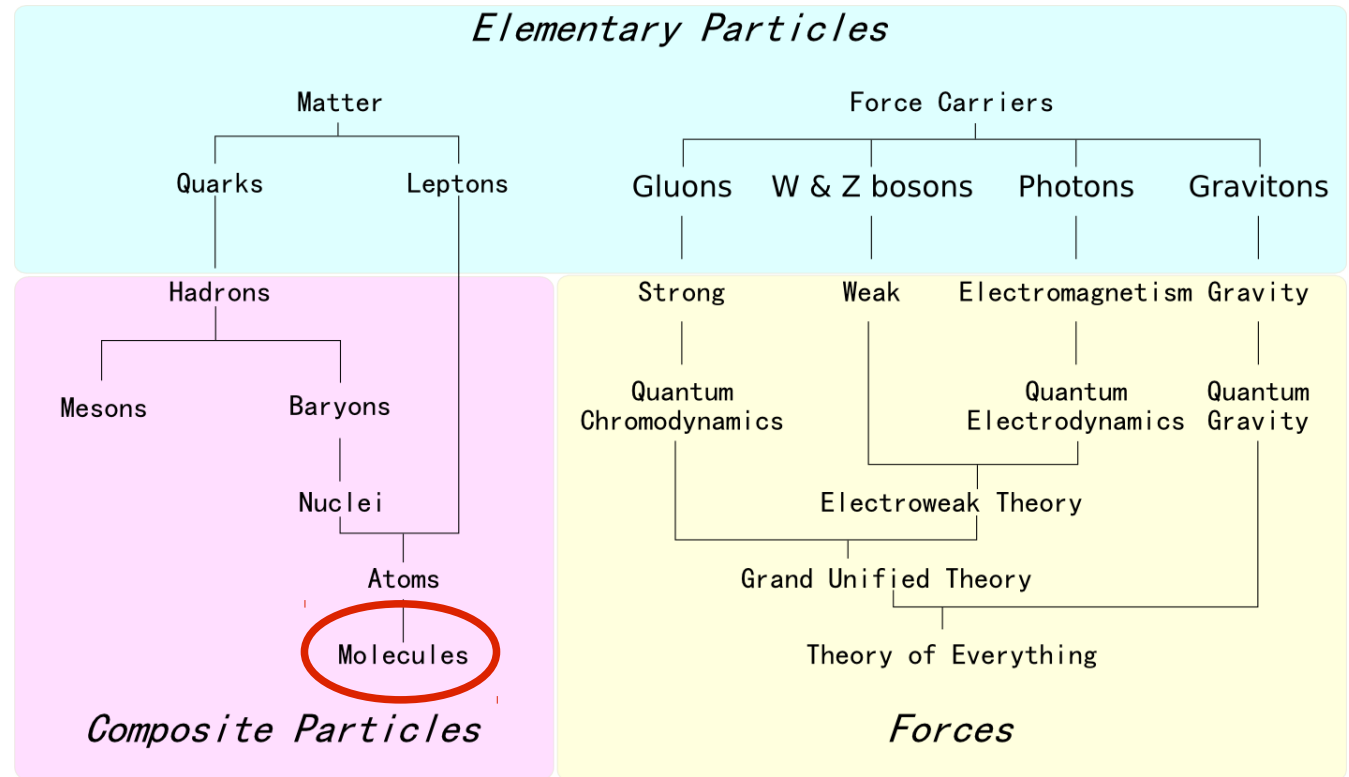
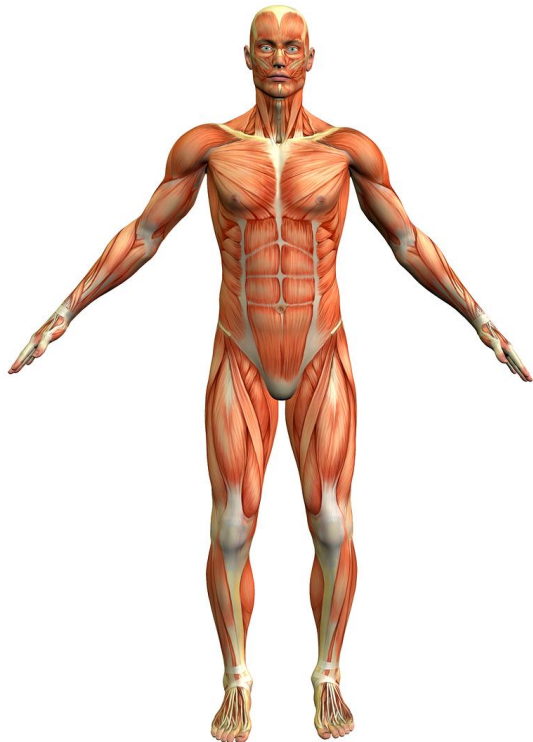
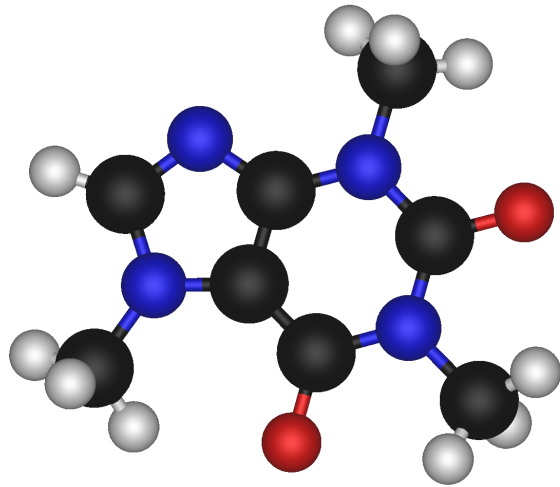
The origin of matter



Molecule:
 n atom-x + m atoms-y, etc.
 Charge = 0

Example: Caffeine = $8x\text{C} + 10x\text{H} + 4x\text{N} + 2x\text{O}$

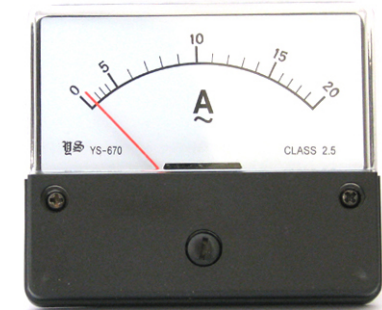
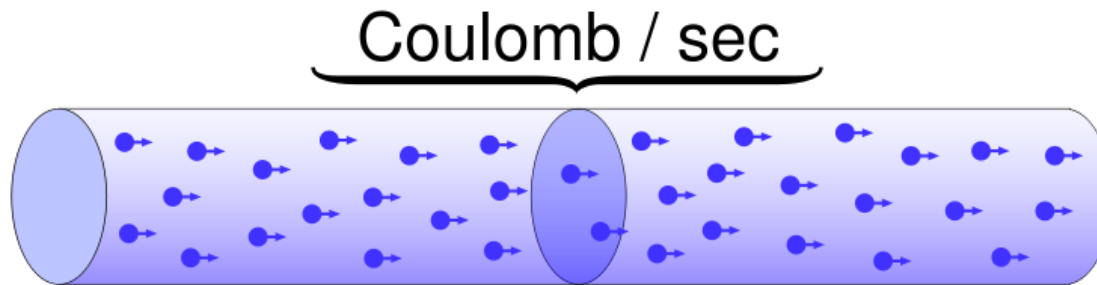
The origin of matter



Matter:
 Combined molecules
 Charge = 0

Example: Human

The origin of current



Current is the passage of charge

- Electrons
- Protons (quarks)

1 unit of charge is $q = 1.6 \times 10^{-19}$ C

1 ampere is by definition 1 coulomb per second ($1 \text{ A} = 1 \text{ C/s}$)

In a typical domestic appliance (fx, vacuum cleaner): about 10^{19} electrons per second!



The magnitude of current

In a typical domestic appliance (fx, vacuum cleaner): about 10^{19} electrons per second!

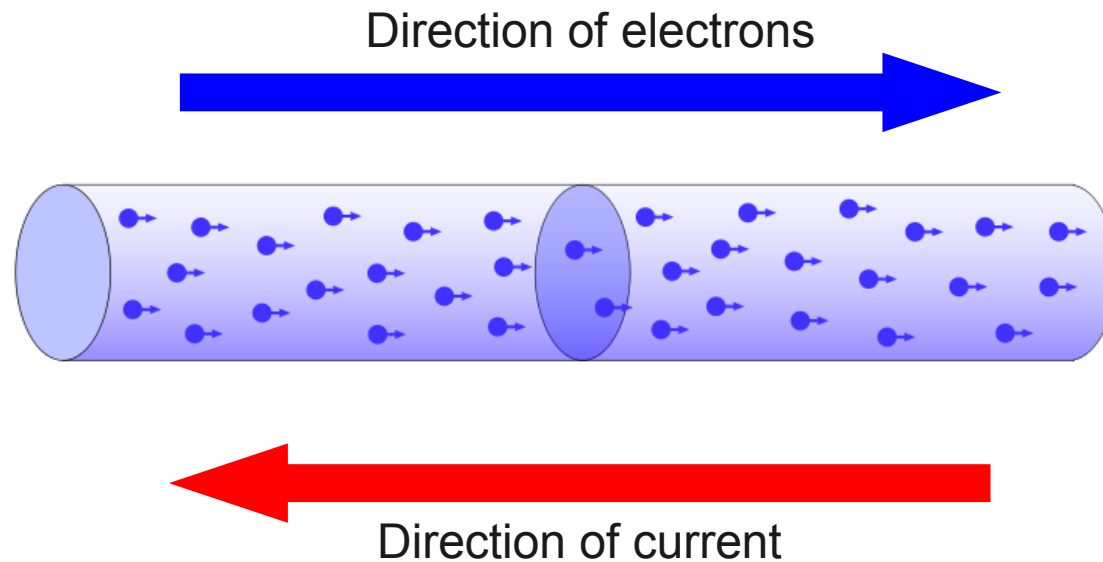


Ontario Highway 401 (busiest road in the world): 430,000 cars/day

In 1 second in a vacuum cleaner pass as many electrons as cars on that highway in **64 billion years!** (Mote: the universe is only 13.7 billion years old)

The sign of current

Current has opposite sign compared to movement of electrons:



$$I = dQ/dt = Q v A$$

Q: charge (C) = $n \times (-q)$!

v: velocity (m/s)

A: area (m²)

n: density of electrons (1/m³)

q: unitary charge (C)

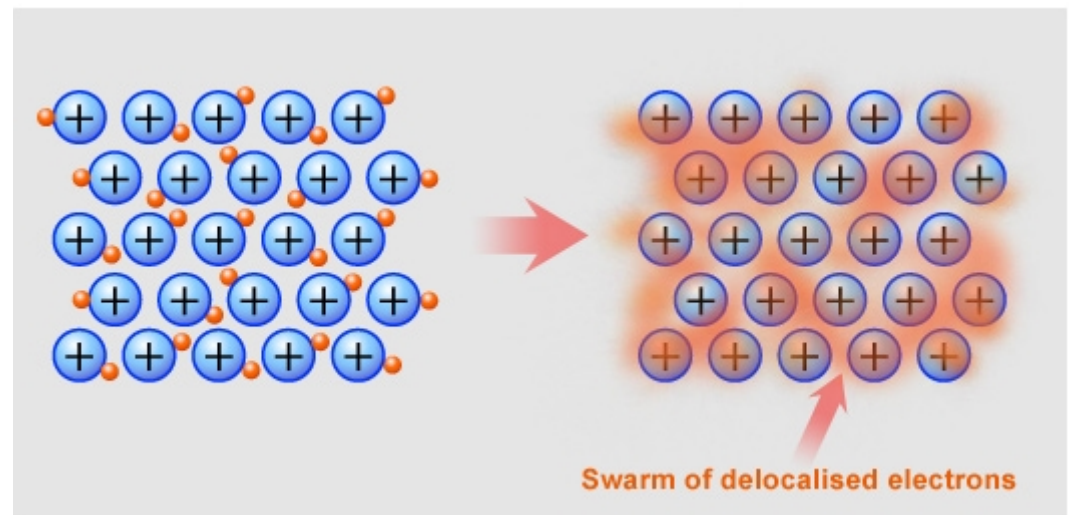
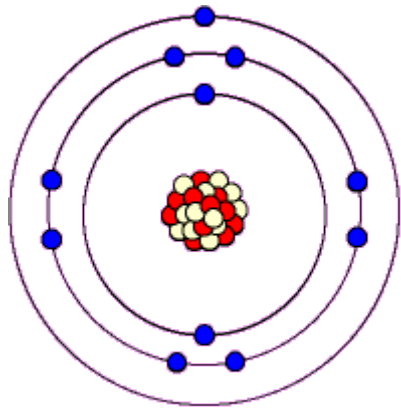
I: current (A = C/s)

Separation of charge. Metals

Objects (atoms, molecules, humans, resistors, capacitors) have **zero net charge**

To have current, positive and negative charge has to be **separated**

1: In **metals** some of the electrons are disconnected from the nuclei and can move freely

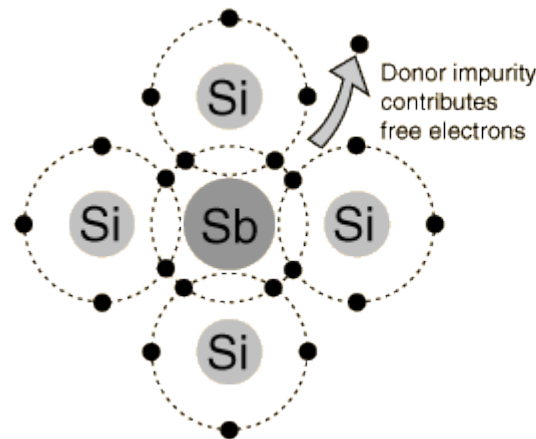
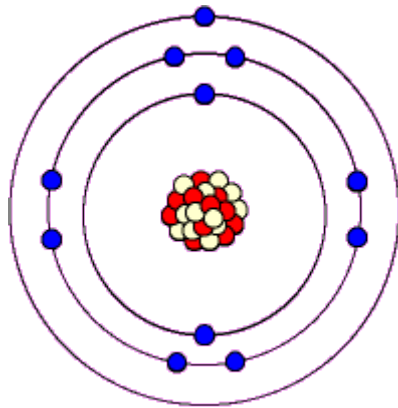


Separation of charge. Semiconductors

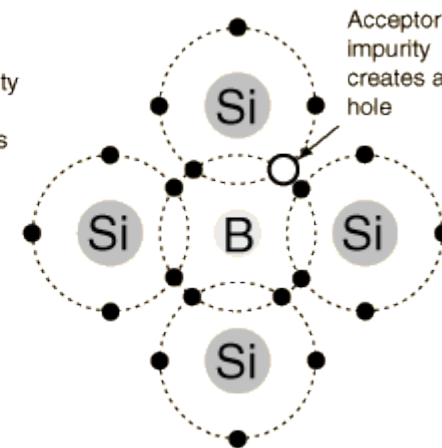
Objects (atoms, molecules, humans, resistors, capacitors) have **zero net charge**

To have current, positive and negative charge has to be **separated**

1: In **semiconductors** some of the atoms have an electron too many or too few for bonding with other atoms



Donor: 1 electron too many

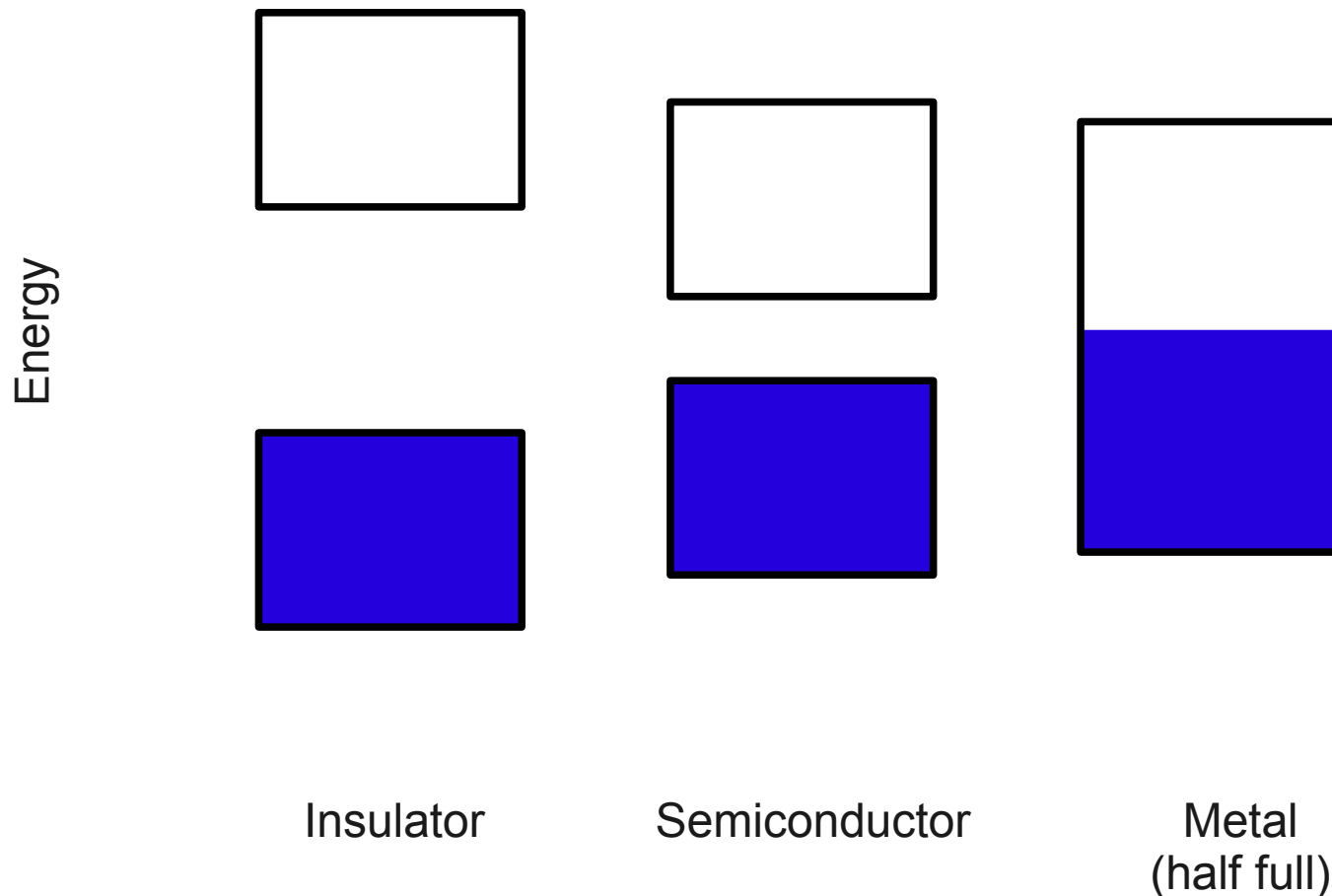


Acceptor: 1 electron too few

Extra electron can be shaken lose from atom to make semiconductor behave like metal

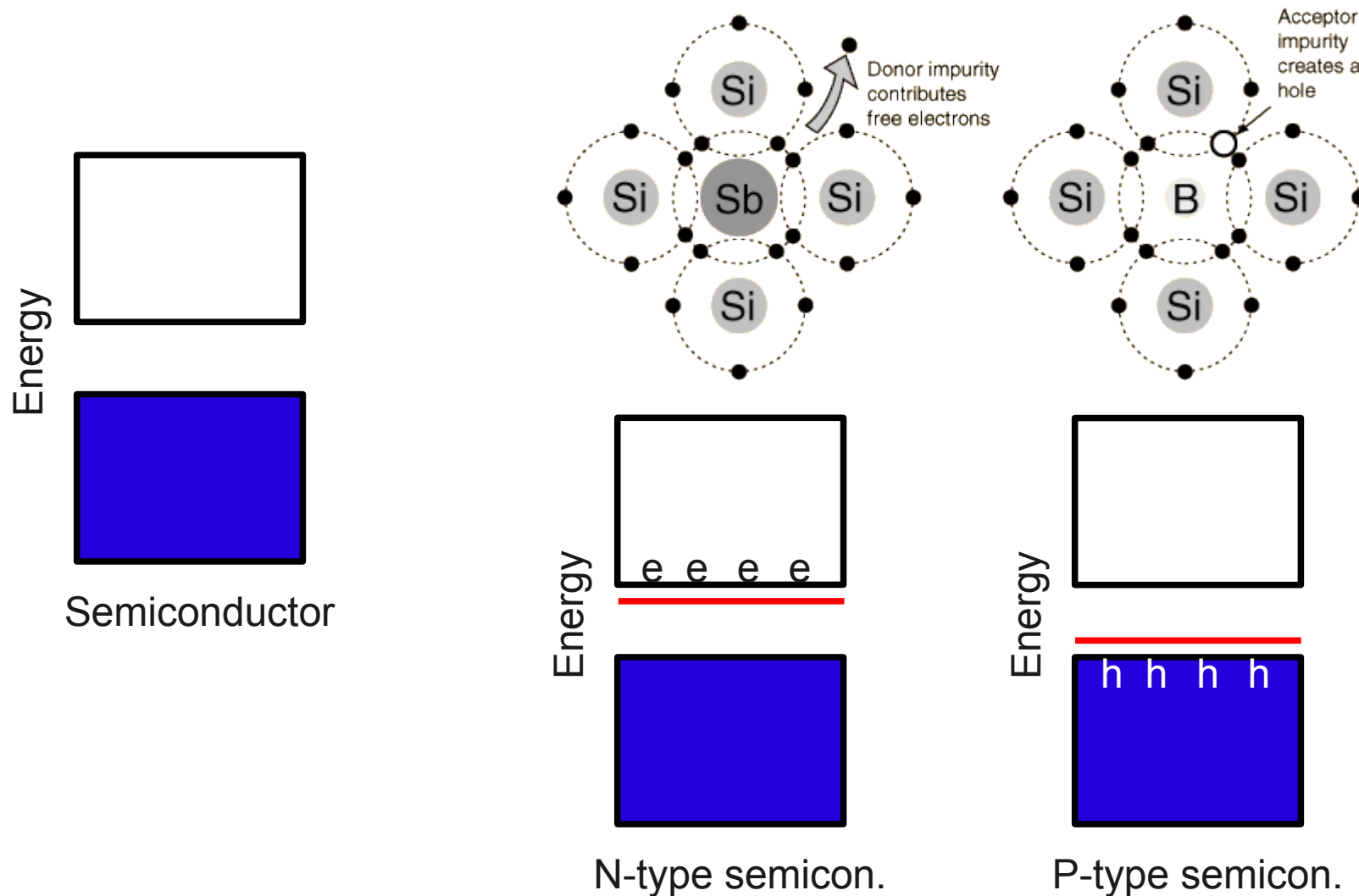
Separation of charge. Band structure

Electrons of materials fill up from lowest energy to highest energy
Not all energies are possible (quantum mechanics. Pauli exclusion principle)



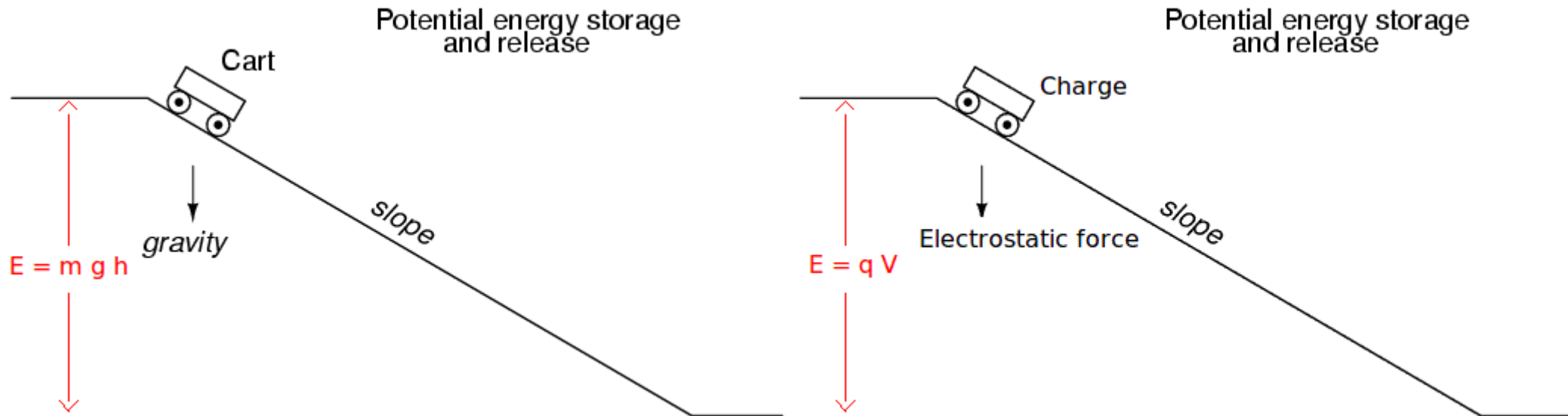
Separation of charge. Band structure

Doping in semiconductors can make charge flow freely



Volt

Now that we know what is current. How to make it happen?



Gravity:

Two objects with mass attract each other. Cart + Earth.

Potential energy:

$$E = m g h$$

Electrostatic interaction:

Two objects with different sign charge attract each other.

Electron + Ion

Potential energy:

$$E = q V$$

Volt

Volt (electrical potential) is like mountains



Volt

It costs energy to separate the charge. Like separating mass (rolling a ball up a hill)

The volt is per definition the energy per unit charge:

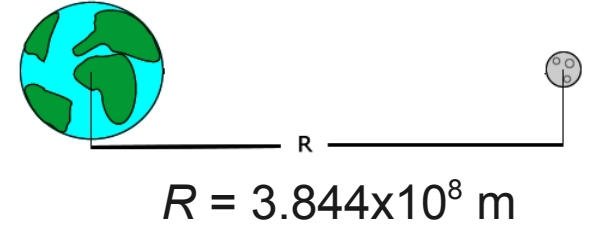
$$1 \text{ volt} = 1 \text{ joule per coulomb}$$



Volt

Force 2 x 1 kg separated by distance Earth-Moon

$$F = G m_1 m_2 / R^2 \quad (G = 6.674 \times 10^{-11} \text{ N m}^2/\text{kg}^2)$$



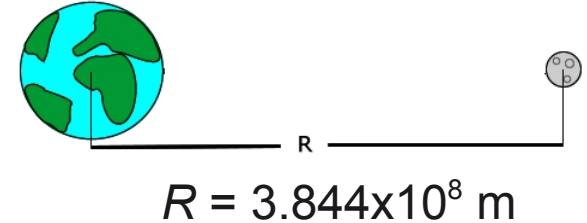
$$F = 4.5 \times 10^{-28} \text{ N}$$



Volt

Force 2 x 1 kg separated by distance Earth-Moon

$$F = G m_1 m_2 / R^2 \quad (G = 6.674 \times 10^{-11} \text{ N m}^2/\text{kg}^2)$$



$$F = 4.5 \times 10^{-28} \text{ N}$$

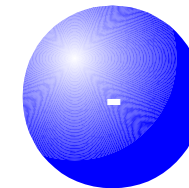


Imagine the two weights stripped of all electrons
(Iron: 26 electrons/atom. Atomic weight: 55.845 u, $u = 1.66 \times 10^{-27} \text{ kg}$)
 $q_1 = q_2 = 4.49 \times 10^7 \text{ C}$

$$F = k q_1 q_2 / R^2 \quad (k = 8.988 \times 10^9 \text{ N m}^2/\text{C}^2)$$



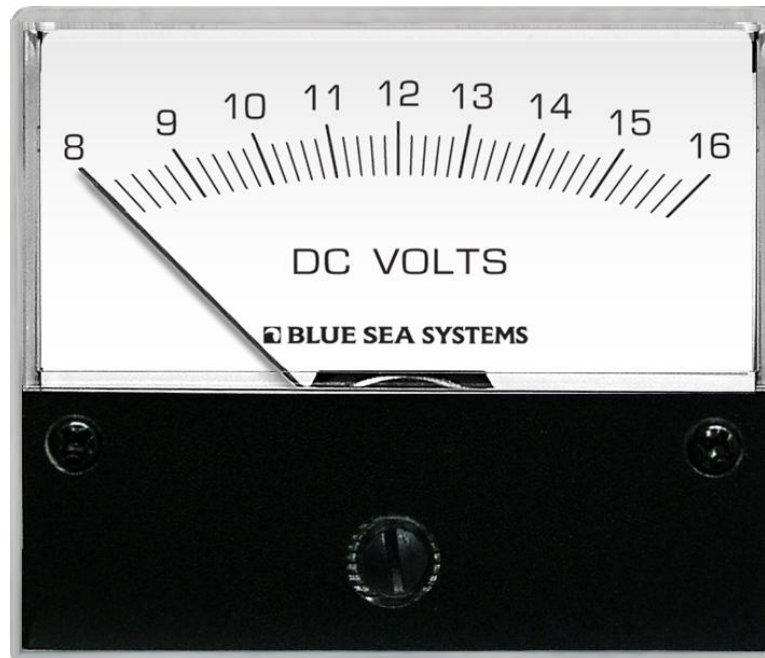
$$F = 1.2 \times 10^8 \text{ N}$$



Force is 35 orders of magnitude larger!

Voltmeter

We can measure the potential energy with a voltmeter

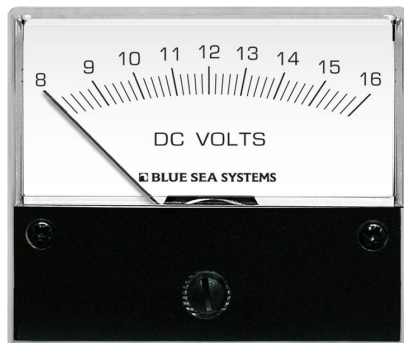


Volt times current is power

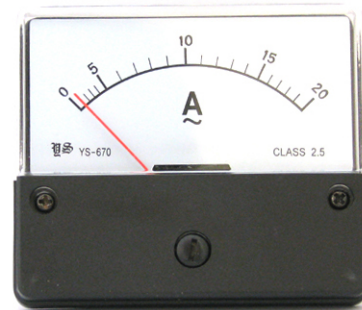
If volt is 'energy per charge' and current is 'charge passing per time', the product of the two is power

volt = joule / coulomb, ampere = coulomb / second

volt x ampere = (J/C)x(C/s) = joule / second = watt



X



=



$$V \times I = P$$

Multimeter

Everything can be measured with a multimeter



Note there is no 'power' meter on a multimeter

kWh

If power is the product of volt and ampere, the integral of power is energy

volt = joule / coulomb, ampere = coulomb / second

volt x ampere = (J/C)x(C/s) = joule / second = watt

energy = power x time

joule = (joule/second) x second

$$P(t) = V(t) \times I(t)$$

$$E = \int P(t)dt = \int V(t)I(t) dt$$



MIEET. The levels of knowledge

